# Synthesis of Isoxazolo[2,3-a]quinoxalines and Pyrrolo[1,2-a]quinoxalines by 1,3-Dipolar Cycloaddition Reaction [1]

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The isoxazolo[2,3-a]quinoxalines 11a,b and pyrrolo[1,2-a]quinoxalines 12a,b were selectively synthesized from the 2-substituted 6-chloroquinoxaline 4-oxides 10a,b. The pyrrolo[1,2-a]quinoxalines 12a,b were clarified to be produced by the ring transformation of the isoxazolo[2,3-a]quinoxalines 11a,b. The pyrrolo[1,2-a]quinoxalines 14a,b were obtained from both 2,6-dichloroquinoxaline 4-oxide 9 and compounds 12a,b.

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There have been many papers on the synthesis of various heterocyclic compounds utilizing the 1,3-dipolar cycloaddition reaction [3], while there have been a few papers concerning the 1,3-dipolar cycloaddition reaction of quinoxaline N-oxides. For example, the reaction of the quinoxaline 1-oxide 1 with dimethyl maleate or N-phenylmaleimide gave the isoxazolo[2,3-a]quinoxalines 4, and the reaction of the quinoxaline 1,4-dioxide 2 with dimethyl acetylenedicarboxylate (DMAD) or N-phenylmaleimide afforded the diisoxazolo[2,3-a:3',2'-c]quinoxalines 5 (Chart 1) [4]. On the other hand, the reaction of the quinoxaline 4-oxide 3 with phenyl isocyanate or benzyne provided the 3-anilinoquinoxaline 6 or 3-(o-hydroxy)phenylquinoxaline 7 via the decarboxylation or isomerization of intermediary condensed quinoxalines, respectively [5]. However, there have been few papers concerning the synthesis of pyrrolo-[1,2-alguinoxalines by the 1,3-dipolar cycloaddition reaction of quinoxaline N-oxides or by the ring transformation of isoxazolo[2,3-a]quinoxalines [6]. In the present in-

Chart 1

vestigation, we found that the 2-substituted 6-chloroquinoxaline 4-oxides 10 were selectively transformed into the isoxazolo[2,3-a]quinoxalines 11 and pyrrolo[1,2-a]quinoxalines 12 (Scheme 1). Moreover, the pyrrolo[1,2-a]quinoxalines 12 were found to be produced by the ring transformation of the isoxazolo[2,3-a]quinoxalines 11. This paper describes the above selective synthesis of 11 and 12 together with a postulated mechanism for the ring transformation of 11 into 12.

The reaction of 2,6-dichloroquinoxaline 8 [7] with m-chloroperbenzoic acid gave 2,6-dichloroquinoxaline 4-oxide 9, whose reaction with piperidine or morpholine afforded 6-chloro-2-(piperidin-1-yl)quinoxaline 4-oxide 10a or 6-chloro-2-(morpholin-4-yl)quinoxaline 4-oxide 10b, respectively. The reaction of 10a or 10b with an equimolar amount of DMAD provided 8-chloro-2,3-bismethoxycarbonyl-4-(piperidin-1-yl)-3aH-isoxazolo[2,3-a]quinoxaline 11a or 8-chloro-2,3-bismethoxycarbonyl-4-(morpholin-4-vl)-3aH-isoxazolo[2,3-a]quinoxaline 11b, respectively. On the other hand, the reaction of 10a or 10b with 2-fold molar amount of DMAD furnished 8-chloro-1,2,3trismethoxycarbonyl-4-(piperidin-1-yl)pyrrolo[1,2-a]quinoxaline 12a or 8-chloro-1,2,3-trismethoxycarbonyl-4-(morpholin-4-yl)pyrrolo[1,2-a]quinoxaline 12b, respectively. The reaction of 11a or 11b with an equimolar amount of DMAD resulted in ring transformation to give 12a or 12b, respectively.

The reaction of **8** with 2-fold molar amount of DMAD provided 4,8-dichloro-1,2,3-trismethoxycarbonylpyrrolo-[1,2-a]quinoxaline **13**, while an attempt was unsuccessful to isolate 2,6-dichloro-2,3-bismethoxycarbonylisoxazolo-[2,3-a]quinoxaline from the reaction of **8** with an equimolar amount of DMAD. The reaction of **13** with piperidine or morpholine did not produce **12a,b**, but resulted in hydrolysis and decarboxylation of the C<sub>1</sub>-ester group to

## Scheme 1

furnish 8-chloro-2,3-bismethoxycarbonyl-4-(piperidin-1-yl)-pyrrolo[1,2-a]quinoxaline 14a or 8-chloro-2,3-bismethoxycarbonyl-4-(morpholin-4-yl)pyrrolo[1,2-a]quinoxaline 14b, respectively. Moisture and secondary amine would promote the formation of intermediates A. Compounds 14a,b were also obtained from 12a,b under a similar reaction condition.

A postulated reaction mechanism for the ring transformation of 11a,b into 12a,b is shown in Scheme 2. The formation of 13 from 9 would also follow this mechanism.

The structural assignment for the above compounds 9-14 was based on the analytical and spectral data. The composition of the isoxazolo[2,3-a]quinoxalines 11a,b was checked by the high resolution mass spectral data, since

11a,b were rather unstable for heating and decomposed in part while recrystallization. The structure of 14a,b was further confirmed by the NOE measurement between C<sub>1</sub>-H

## Chart 2

NOE Data for Compounds 14a,b.

and C<sub>9</sub>-H (Chart 2). Namely, the radiation at the C<sub>1</sub>-H or C<sub>9</sub>-H proton signal showed 10-18.5% NOE to the C<sub>9</sub>-H or C<sub>1</sub>-H proton signal, respectively.

#### EXPERIMENTAL

All melting points were determined on a Ishii melting point apparatus and are uncorrected. The ir spectra (potassium bromide) were recorded with a JASCO IRA-1 spectrophotometer. The nmr spectra were measured in deuteriodimethyl sulfoxide with a VXR-300 spectrometer at 300 MHz. Chemical shifts are given in the  $\delta$  scale. The mass spectra (ms) were determined with a JEOL JMS-01S spectrometer. Elemental analyses were performed on a Perkin-Elmer 240B instrument.

## 2,6-Dichloroquinoxaline 4-Oxide 9.

A solution of **8** (20 g, 0.10 mole) and m-chloroperbenzoic acid (27.2 g, 1.1 equivalents) in chloroform (500 ml) was refluxed on a boiling water bath for 10 hours. Removal of the solvent in vacuo gave crystals, which were triturated with saturated sodium bicarbonate solution to exclude m-chlorobenzoic acid and residual m-chloroperbenzoic acid. The crystals were collected by suction filtration and recrystallized from N,N-dimethylformamide/ethanol provided pale yellow needles **9** (15.5 g, 72%); mp 176-178°; ir:  $\nu$  cm<sup>-1</sup> 3060, 1595, 1228; ms: m/z 215 (M\*), 217 (M\*+2); pmr: 8.94 (s, 1H, C<sub>3</sub>-H), 8.35 (d, J = 2.5 Hz, 1H, C<sub>5</sub>-H), 8.07 (d, J = 9.0 Hz, 1H, C<sub>5</sub>-H), 7.83 (dd, J = 2.5 Hz, J = 9.0 Hz, 1H, C<sub>7</sub>-H).

Anal. Calcd. for C<sub>0</sub>H<sub>4</sub>Cl<sub>2</sub>N<sub>2</sub>O: C, 44.68; H, 1.87; Cl, 32.98; N, 13.03. Found: C, 44.80; H, 1.89; Cl, 32.95; N, 12.84.

6-Chloro-2-(piperidin-1-yl)quinoxaline 4-Oxide 10a and 6-Chloro-2-(morpholin-4-yl)quinoxaline 4-Oxide 10b.

A solution of 9 (10 g, 46.5 mmoles) and piperidine (5.94 g, 69.75 mmoles) or morpholine (6.08 g, 69.75 mmoles) in N,N-dimethylformamide (300 ml) was refluxed in an oil bath for 3 hours. Removal of the solvent *in vacuo* gave yellow crystals 10a or 10b, which were collected by suction filtration. Recrystallization from N,N-dimethylformamide/ethanol afforded yellow needles 10a (7.24 g, 59%) or 10b (8.53 g, 69%).

Compound 10a had mp 165-166°; ir:  $\nu$  cm<sup>-1</sup> 3070, 2910, 1572, 1215; ms: m/z 263 (M\*), 265 (M\* + 2); pmr: 8.59 (s, 1H, C<sub>3</sub>-H), 8.14 (d, J = 2.0 Hz, 1H, C<sub>5</sub>-H), 7.66-7.55 (m, 2H, C<sub>7</sub>-H and C<sub>8</sub>-H), 3.68 (t, J = 4.5 Hz, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>), 1.69-1.51 (m, 6H, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>). Anal. Calcd. for C<sub>13</sub>H<sub>14</sub>ClN<sub>3</sub>O: C, 59.21; H, 5.35; Cl, 13.44; N, 15.93. Found: C, 59.30; H, 5.48; Cl, 13.23; N, 15.65.

Compound 10b had mp 152-153°; ir:  $\nu$  cm<sup>-1</sup> 3070, 2940, 1570, 1220; ms: m/z 265 (M\*), 267 (M\*+2); pmr: 8.64 (s, 1H, C<sub>3</sub>-H), 8.17 (d, J = 2.0 Hz, 1H, C<sub>5</sub>-H), 7.71-7.58 (m, 2H, C<sub>7</sub>-H and C<sub>8</sub>-H), 3.75-3.62 (m, 8H, morpholine CH<sub>2</sub>).

Anal. Calcd. for C<sub>12</sub>H<sub>12</sub>ClN<sub>3</sub>O<sub>2</sub>: C, 54.25; H, 4.55; Cl, 13.34; N, 15.82. Found: C, 54.24; H, 4.52; Cl, 13.49; N, 15.65.

8-Chloro-2,3-bismethoxycarbonyl-4-(piperidin-1-yl)-3aH-isoxazolo-[2,3-a]quinoxaline 11a.

A suspension of 10a (2 g, 7.6 mmoles) and dimethyl acetylenedicarboxylate (1.2 g, 8.36 mmoles) in cyclohexane (200 ml) was refluxed on a boiling water bath for 1 hour to precipitate red needles 11a, which were collected by suction filtration and then triturated with ethanol (drying: below 80° in vacuo) (3.02 g, 98%), mp 224-225°; ir:  $\nu$  cm<sup>-1</sup> 3050, 2920, 1730, 1655, 1595; ms:

m/z 405 (M\*), 407 (M\* + 2); pmr: 9.35 (s, 1H), 7.90 (s, 2H), 7.70 (s, 1H) (C<sub>3a</sub>-H, C<sub>6</sub>-H, C<sub>7</sub>-H and C<sub>9</sub>-H), 3.85 (s, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>), 3.79 (s, 3H, CH<sub>3</sub>), 3.50 (s, 3H, CH<sub>3</sub>), 1.65 (s, 6H, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>).

Anal. Calcd. for  $C_{19}H_{20}ClN_3O_5$ : 405.109 (M<sup>+</sup>), 407.106 (M<sup>+</sup> + 2). Found: 405.110 (M<sup>+</sup>), 407.105 (M<sup>+</sup> + 2).

8-Chloro-2,3-bismethoxycarbonyl-4-(morpholin-4-yl)-3aH-isoxazolo[2,3-a]quinoxaline 11b.

A suspension of **10b** (2 g, 7.6 mmoles) and dimethyl acetylenedicarboxylate (1.19 g, 8.36 mmoles) in dioxane (60 ml) was refluxed in an oil bath for 1 hour to precipitate red needles **11b**, which were collected by suction filtration and then triturated with ethanol (drying: below 50° in vacuo) (2.97 g, 96%), mp 238-239°; ir:  $\nu$  cm<sup>-1</sup> 3050, 2940, 1730, 1658, 1595; ms: m/z 407 (M\*), 409 (M\*+2); pmr: 9.35 (s, 1H), 7.92 (s, 2H), 7.73 (s, 1H) (C<sub>3a</sub>-H, C<sub>6</sub>-H, C<sub>7</sub>-H and C<sub>9</sub>-H), 3.85 (s, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>), 3.78 (s, 3H, CH<sub>3</sub>), 3.58 (s, 3H, CH<sub>3</sub>), 3.52 (s, 4H, CH<sub>2</sub>-O-CH<sub>2</sub>).

Anal. Calcd. for  $C_{18}H_{18}ClN_3O_6$ : 407.088 (M\*), 409.085 (M\* + 2). Found: 407.085 (M\*), 409.082 (M\* + 2).

8-Chloro-1,2,3-trismethoxycarbonyl-4-(piperidin-1-yl)pyrrolo-[1,2-a]quinoxaline 12a and 8-Chloro-1,2,3-trismethoxycarbonyl-4-(morpholin-4-yl)pyrrolo[1,2-a]quinoxaline 12b.

A solution of 10a (2 g, 7.6 mmoles) or 10b (2 g, 7.6 mmoles) and dimethyl acetylenedicarboxylate (2.35 g, 16.7 mmoles) in dioxane (60 ml) was refluxed in an oil bath for 6 hours. Removal of the solvent in vacuo afforded crystals, which were triturated with ethanol/hexane and then collected by suction filtration. Recrystallization from N,N-dimethylformamide/water provided colorless needles 12a (1.06 g, 30%) or 12b (0.94 g, 27%).

Compound 12a had mp 152-153°; ir:  $\nu$  cm<sup>-1</sup> 3120, 2940, 1730, 1710, 1598; ms: m/z 459 (M<sup>+</sup>), 461 (M<sup>+</sup>+2); pmr: 7.69 (d, J = 2.0 Hz, 1H, C<sub>9</sub>-H), 7.68 (d, J = 9.0 Hz, 1H, C<sub>6</sub>-H), 7.53 (dd, J = 2.0 Hz, J = 9.0 Hz, 1H, C<sub>7</sub>-H), 4.02 (s, 3H, CH<sub>3</sub>), 3.88 (s, 3H, CH<sub>3</sub>), 3.85 (s, 3H, CH<sub>3</sub>), 3.28 (s, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>), 1.58 (s, 6H, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>3</sub>).

Anal. Calcd. for C<sub>22</sub>H<sub>22</sub>ClN<sub>3</sub>O<sub>6</sub>: C, 57.46; H, 4.82; Cl, 7.71; N, 9.14. Found: C, 57.53; H, 4.85; Cl, 8.01; N, 9.25.

Compound 12b had mp 164-166°; ir:  $\nu$  cm<sup>-1</sup> 3110, 2950, 1725, 1710, 1598; ms: m/z 461 (M\*), 463 (M\*+2); pmr: 7.72 (d, J = 9.0 Hz, 1H, C<sub>6</sub>-H), 7.71 (d, J = 2.5 Hz, 1H, C<sub>9</sub>-H), 7.57 (dd, J = 9.0 Hz, J = 2.5 Hz, 1H, C<sub>7</sub>-H), 4.04 (s, 3H, CH<sub>3</sub>), 3.89 (s, 3H, CH<sub>3</sub>), 3.86 (s, 3H, CH<sub>3</sub>), 3.70 (t, J = 4.5 Hz, 4H, CH<sub>2</sub>-O-CH<sub>2</sub>), 3.29 (t, J = 4.5 Hz, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>).

Anal. Calcd. for C<sub>21</sub>H<sub>20</sub>ClN<sub>3</sub>O<sub>7</sub>: C, 54.61; H, 4.37; Cl, 7.68; N, 9.10. Found: C, 54.56; H, 4.39; Cl, 7.68; N, 9.22.

## Ring Transformation of 11a,b into 12,b.

A solution of **11a** or **11b** (2 g) and dimethyl acetylenedicarboxylate (0.77 g, 1.1-fold) in dioxane (60 ml) was refluxed in an oil bath for 7 hours. Removal of the solvent *in vacuo* afforded crystals, which were triturated with ethanol and then collected by suction filtration. Recrystallization from *N,N*-dimethylformamide/water gave colorless needles **12a** (0.45 g, 20%) or **12b** (0.75 g. 33%).

4,8-Dichloro-1,2,3-trismethoxycarbonylpyrrolo[1,2-a]quinoxaline 13.

A solution of 9 (5 g, 23.26 mmoles) and dimethyl acetylenedicarboxylate (7.26 g, 51.17 mmoles) in dioxane (200 ml) was refluxed in an oil bath for 4 hours. Evaporation of the solvent *in vacuo* gave colorless crystals, which were triturated with ethanol/hexane

and then collected by suction filtration (2.1 g, 22%). Recrystallization from ethanol/hexane provided colorless needles, mp 200-202°; ir:  $\nu$  cm<sup>-1</sup> 2950, 1730, 1680, 1610; ms: m/z 410 (M<sup>+</sup>), 412 (M<sup>+</sup> + 2); pmr: 7.96 (d, J = 8.5 Hz, 1H, C<sub>6</sub>-H), 7.74 (d, J = 8.5 Hz, J = 2.0 Hz, 1H, C<sub>7</sub>-H), 7.70 (d, J = 2.0 Hz, 1H, C<sub>9</sub>-H), 4.09 (s, 3H, CH<sub>3</sub>), 3.94 (s, 3H, CH<sub>3</sub>), 3.87 (s, 3H, CH<sub>3</sub>).

Anal. Calcd. for C<sub>17</sub>H<sub>12</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>6</sub>: C, 49.66; H, 2.94; Cl, 17.24; N, 6.81. Found: C. 49.88; H. 2.92; Cl. 17.08; N, 6.88.

8-Chloro-2,3-bismethoxycarbonyl-4-(piperidin-1-yl)pyrrolo[1,2-a]-quinoxaline **14a** and 8-Chloro-2,3-bismethoxycarbonyl-4-(morpholin-4-yl)pyrrolo[1,2-a]quinoxaline **14b**.

A solution of 13 (1 g, 2.43 mmoles) and piperidine (620 mg, 7.29 mmoles) or morpholine (634 mg, 7.29 mmoles) in N,N-dimethylformamide (30 ml) was refluxed in an oil bath for 3 hours. Evaporation of the solvent in vacuo gave an oily substance, which was dissolved in hot ethanol. Cooling of the solution to room temperature precipitated colorless needles 14a or 14b, which were collected by suction filtration, yield, 14a (340 mg, 35%), 14b (460 mg, 47%). Recrystallization from N,N-dimethylformamide/ethanol afforded colorless needles 14a and colorless prisms 14b.

Compound 14a had mp 214-215°; ir:  $\nu$  cm<sup>-1</sup> 1735, 1690; ms: m/z 401 (M\*), 403 (M\*+2); pmr: 9.02 (s, 1H, C<sub>1</sub>-H), 8.57 (d, J = 2.1 Hz, 1H, C<sub>9</sub>-H), 7.66 (d, J = 8.5 Hz, 1H, C<sub>6</sub>-H), 7.45 (dd, J = 8.5 Hz, J = 2.1 Hz, 1H, C<sub>7</sub>-H), 3.88 (s, 3H, CH<sub>3</sub>), 3.84 (s, 3H, CH<sub>3</sub>), 3.24 (s, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>), 1.59 (s, 6H, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>).

Anal. Caled. for C<sub>20</sub>H<sub>20</sub>ClN<sub>3</sub>O<sub>4</sub>: C, 59.78; H, 5.02; Cl, 8.82; N, 10.46. Found: C, 59.57; H, 5.05; Cl, 8.77; N, 10.45.

Compound 14b had mp 225-226°; ir:  $\nu$  cm<sup>-1</sup> 1730, 1710; ms: m/z 403 (M<sup>+</sup>), 405 (M<sup>+</sup>+2); pmr: 9.08 (s, 1H, C<sub>1</sub>-H), 8.62 (d, J = 2.3 Hz, 1H, C<sub>9</sub>-H), 7.66 (d, J = 8.5 Hz, 1H, C<sub>6</sub>-H), 7.49 (dd, J = 8.5 Hz, J = 2.3 Hz, 1H, C<sub>7</sub>-H), 3.88 (s, 3H, CH<sub>3</sub>), 3.84 (s, 3H, CH<sub>3</sub>), 3.70 (t, J = 4.0 Hz, 4H, CH<sub>2</sub>-O-CH<sub>2</sub>), 3.26 (t, J = 4.0 Hz, 4H, CH<sub>2</sub>-N-CH<sub>2</sub>).

Anal. Calcd. for C<sub>19</sub>H<sub>18</sub>ClN<sub>3</sub>O<sub>5</sub>: C, 56.51; H, 4.49: Cl, 8.78; N, 10.41. Found: C, 56.46; H, 4.53; Cl, 8.85; N, 10.39.

Synthesis of 14a,b from 12a,b.

A suspension of 12a (500 mg, 1.08 mmole) and piperidine (277 mg, 3.26 mmoles) in N,N-dimethylformamide (30 ml) was refluxed in an oil bath for 3 hours. Evaporation of the solvent in vacuo gave an oily substance, which was dissolved in hot ethanol/water. Cooling of the solution to room temperature precipitated colorless needles 14a, which were collected by suction filtration (370 mg, 79%).

The reaction of **12b** (500 mg, 1.08 mmole) and morpholine (284 mg, 3.25 mmoles) in *N*,*N*-dimethylformamide (30 ml) under a similar condition to the above provided **14b** as colorless prisms (400 mg, 92%).

#### REFERENCES AND NOTES

- [1] Preliminary report: H. S. Kim, Y. Kurasawa and A. Takada, J. Heterocyclic Chem., 26, 871 (1989).
- [2] Present address: Department of Chemistry, Teacher's College, Hyosung Women's University, Gyongsan 713-900, Korea.
- [3] A. Padwa, 1,3-Dipolar Cycloaddition Chemistry, Vols I, II, E. C. Taylor and A. Weissberger, eds, John Wiley and Sons, New York, Chichester, Brisbane, Toronto, Singapore, 1984; D. P. Curran, Advances in Cycloaddition, Vol I, D. P. Curran, ed, JAI Press Inc., Connecticut, London, 1988, and references cited therein.
- [4] M. Ungreanu, I. Druta and I. Zugravescu, An. Stiint. Univ. "Al. I. Cuza" Iasi, Sect. 1c, 20, 29 (1974); Chem. Abstr., 82, 125351q (1975).
- [5] J. C. Mason and G. Tennant, J. Chem. Soc., Chem. Commun., 218 (1972).
- [6] G. W. H. Cheeseman and R. F. Cookson, The Chemistry of Heterocyclic Compounds. Condensed Pyrazines, A. Weissberger and E. C. Taylor, eds, John Wiley and Sons, New York, Chichester, Brisbane, Toronto, 1977, pp 598-622, and references cited therein.
- [7] K. Makino, G. Sakata, K. Morimoto and Y. Ochiai, Heterocycles, 23, 2025 (1985).